HENRY VON OESEN AND ASSOCIATES CONSULTING ENGINEERS AND PLANNERS

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WILMINGTON, NORTH CAROLINA 28402

March 6, 1979

Atlantic Division Naval Facilities Engineering Command Norfolk, Virginia 23511

Attn: Mr. M. L. Bryant, P. E. Re: Contract N62470-78-B-4600

Gentlemen:

We have studied the well water quality at Tarrawa Terrace and Montford Point at Camp Lejeune. Recent analyses of the water from each well presently in use were obtained from the Base Utility Division. These analyses were compared to the applicable requirements of the Safe Drinking Water Act, Bureau of Medicine and Surgery, and the State of North Carolina.

Although complete analyses of all wells were not available, there is sufficient overlapping and data from other sources to give indications of water quality in the area. Copies of the available analyses are attached.

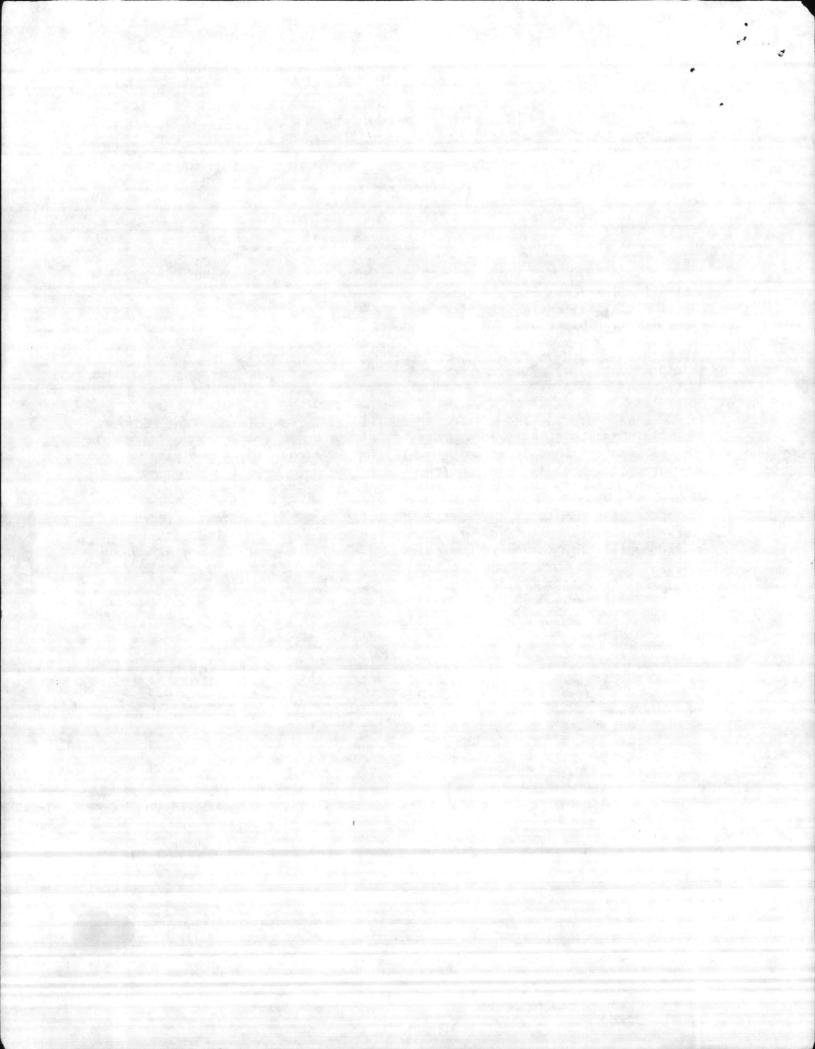
The maximum contaminant level for inorganic chemicals of the Safe Drinking Water Act are as follows:

Arsenic	0.05 mg/l
Barium	1.0
Cadmi um	. 0.01
Chromium	0.05
Lead	0.05
Mercury	0.002
Selenium	0.01
Silver	0.05
Fluoride	1.4 to 2.4 (depending on temperature)
Nitrate	10.0

The State of North Carolina has the following additional maximum contaminant levels for other chemicals and requires treatment to remove amounts in excess:

Iron	0.3
Manganese	0.05

The State also has the following recommended limits for other chemical substances:



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Chloride	250
Copper	1.0
Phenols	0.001
Sulfate	250
Total dissolved solids	500
Zinc	5

All of the wells presently in use meet the Safe Drinking Water Act requirements. They also meet the state recommended limits for other chemical substances. However, the iron content in every well is significantly in excess of the state maximum. Manganese is also slightly high in three wells. Although there are no state or federal limits for hardness, DM-5 recommends treatment of water with hardness in excess of 150. Hardness of the present wells ranges from 164 to 320, and it is the present practice at Camp Lejeune to soften the water. The hardness is virtually all calcium bicarbonate.

Therefore, the water needs treatment for removal of significant amounts of iron and calcium bicarbonate hardness. There are two basic types of treatment that we would recommend for this service, as indicated on the attached sketches.

1. Aeration - filtration - ion exchange

The aerator oxidizes the dissolved ferrous state iron (and manganese) to the ferric state, forming a floc which is settled and filtered out. About 75% of the water will pass through zeolite softeners, and about 25% bypassed to provide a residual hardness of 75 mg/l. There will be the usual applications of chlorine, fluoride, and possibly chemical feeds for flocculation and pH adjustment.

2. Cold lime process

Lime is injected into the raw water as it enters the lime contact tank. Through a catalytic precipitation process, the hardness and iron are precipitated and partially removed by coating on a sand media in the contact tank. The remaining hardness and iron floc are removed in the filter. The rate of lime feed is set to provide a residual hardness of 75 mg/l. Recarbonation may be required if the leaving pH is above 9. The usual applications of chlorine and fluoride will be made.

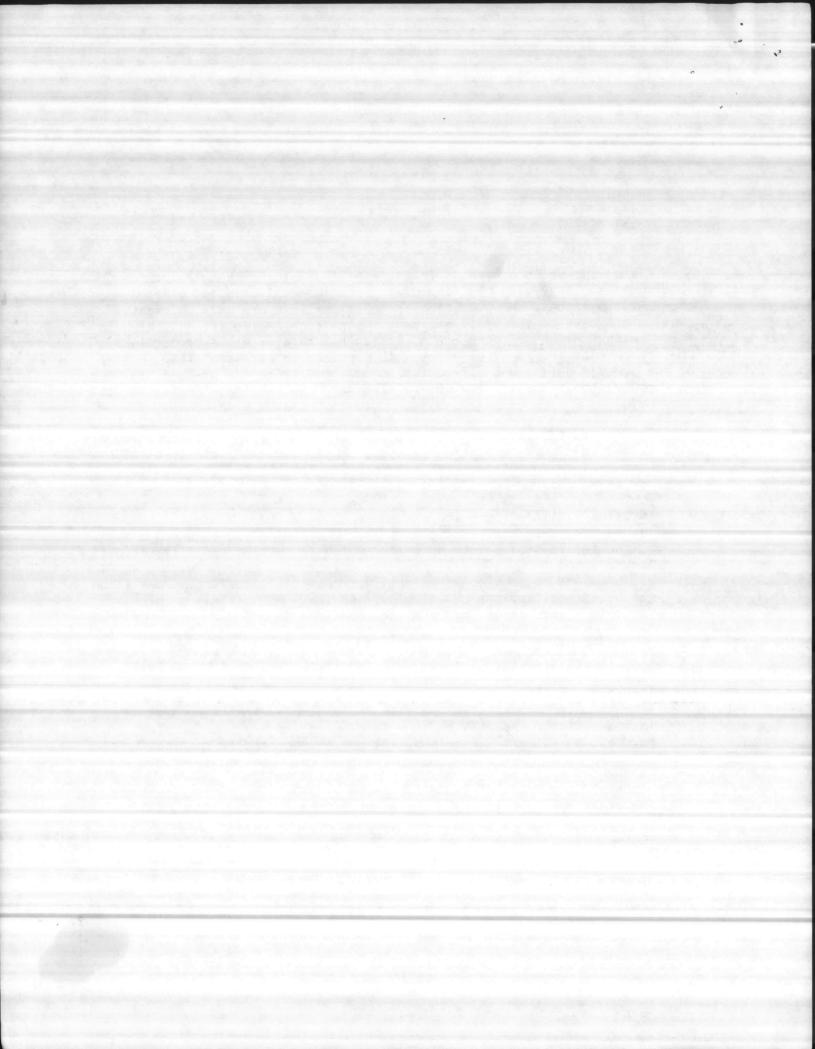
The final report will discuss the merits and economics of each treatment method.

If you have any questions regarding this information, please let us know.

Very truly yours,

HENRY VON OESEN AND ASSOCIATES, INC. eun James R. Benson, Jr., P. E.

JRB/GGB Enc.



	142 M-THE	M-168	M-197	M-628	M-629	M-630			•
Parameter	HANDINOKT XEXXIXXIX	MONATANORIX XEOKIXIXXX XEOKIXIXXX	CXXXXIP CXEXIXCHEXR	XXMMASSINIX	WOXISINOX SHOWIEK	XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX	RIFLE RANGE	HOLCOMB BLVD	NEW RIVER
РН	7.4	7.3	7.3	7.4	7.3	7.4			
PHENOLTHALEIN ALKALINITY	0	0	0	0	0	0			
METHYL ORANGE ALKALINITY HARDNESS	204	230	250	228	178	212			
CARDONALSS AS CaCO3	190	220	320	220	180	308			
BICARBONATES AS CaCO3									
CHLORIDES AS CL	8	14	162	10	12	108			
HARDNESS AS CaCO3		1							
LRON AS Fe	.83	.95	2.05	.85	4.50	1.16		5 - 14 - 14	
MANGANESE XXIXMXXHKOSHHAXE	.03	.03	.04	.03	.06	.03			
CALCIUM XERUQXBHOGBHARE	85	87	125	88	65	115	-		
MAGNESIUM ASEACXEHOGEHASTEC	.1.22	2.31	4.16	1.92	1.68	2.35			
LUORIDE		the inerties in	a des			. 2.35			
HLORINE RESIDUAL									

NOTE: All results reported in parts per million unless otherwise noted except for pH, temperature, and specific conductance. One liter of potable water is assumed to weigh one kilogram.

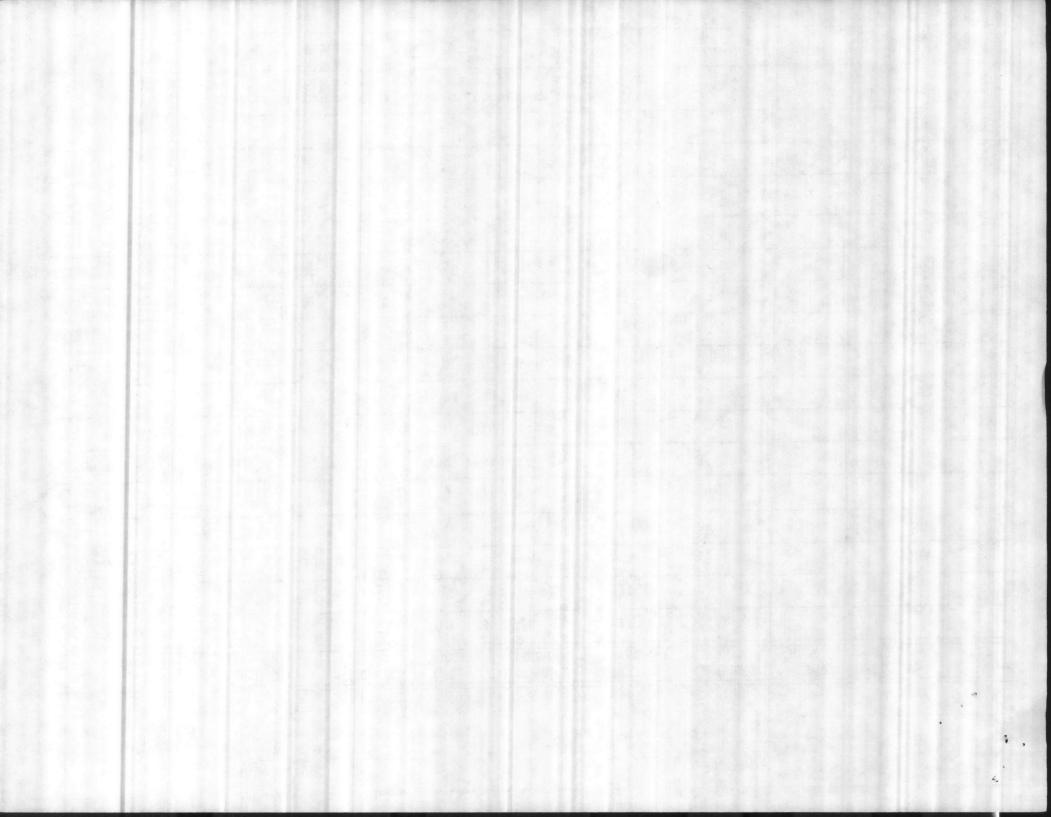
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LABORATORY ANALYSIS BY:

DATE OF ANALYSIS:

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CHEMICAL ANALYSIS - WATER MCBCL 11330/3 (REV 8-74)

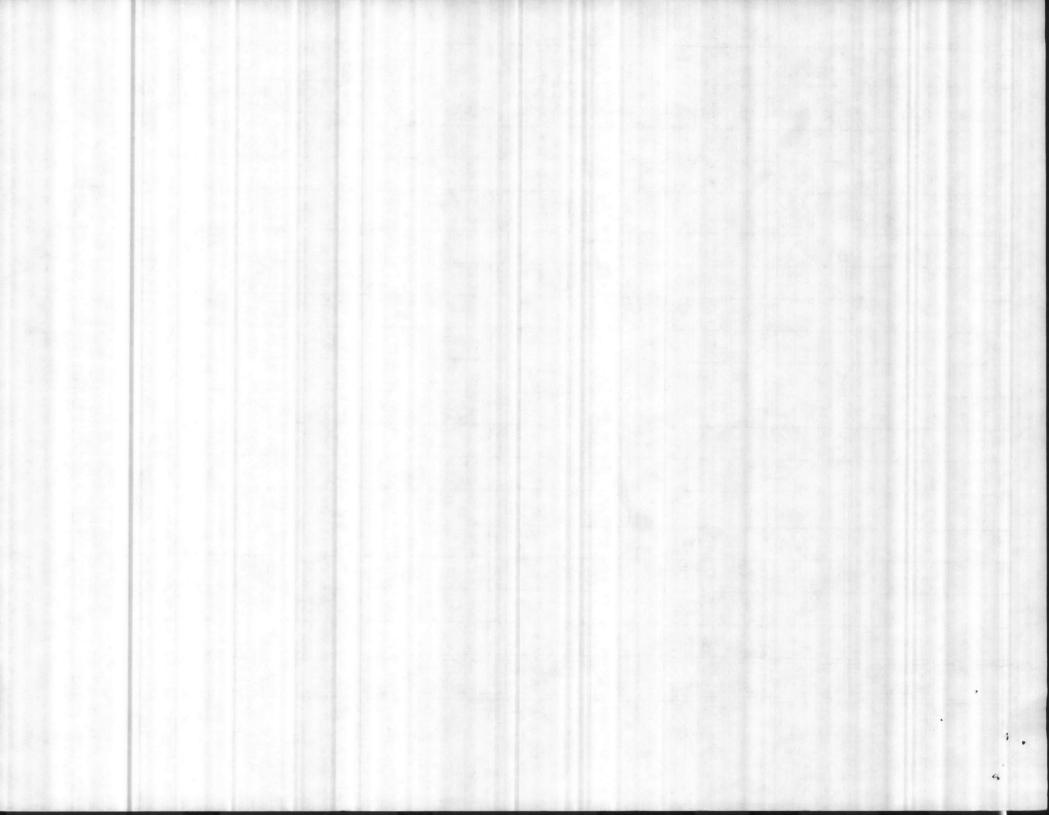
TARAWA TERRACE

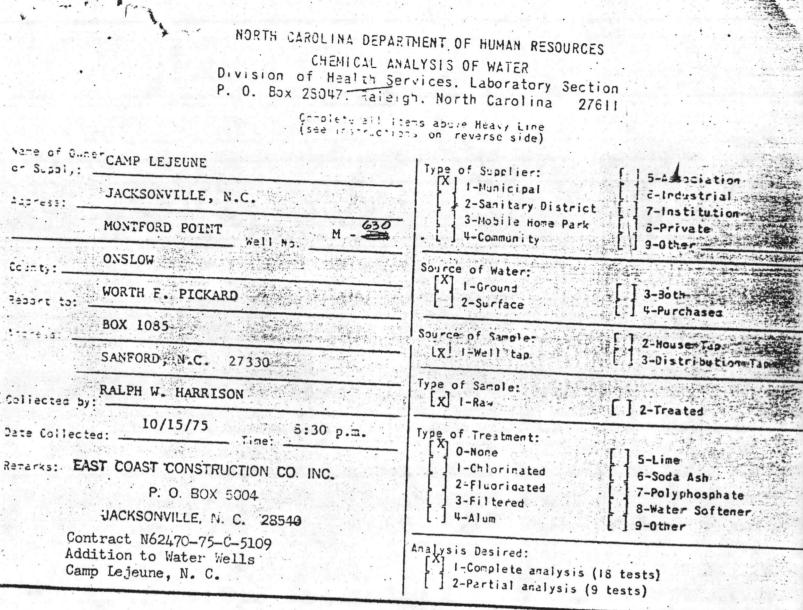
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	POXEME	CHEXIXXEEX	DAXAANAA DEXERAXAE	XHEXXXXHX	XXOOKIXHODEE XBAX	RAKARA	XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX	NEW XRXXER
7.4	7.3	7.2	7.4	7.5	7.7			
0	0	0	0	0	0			
228	224	204	198	180	172			
220	220	256	216	200	164			
	10							
14	16	12	8	12	2			
						de la		
.93	12.0	5.20	1.0	. 53	1.25			
.02	.10	.06	.03	.01	.03			
103	85	106	80	81	67			
2.26	2.2	2.72	1.82	1.40	1.79	•		
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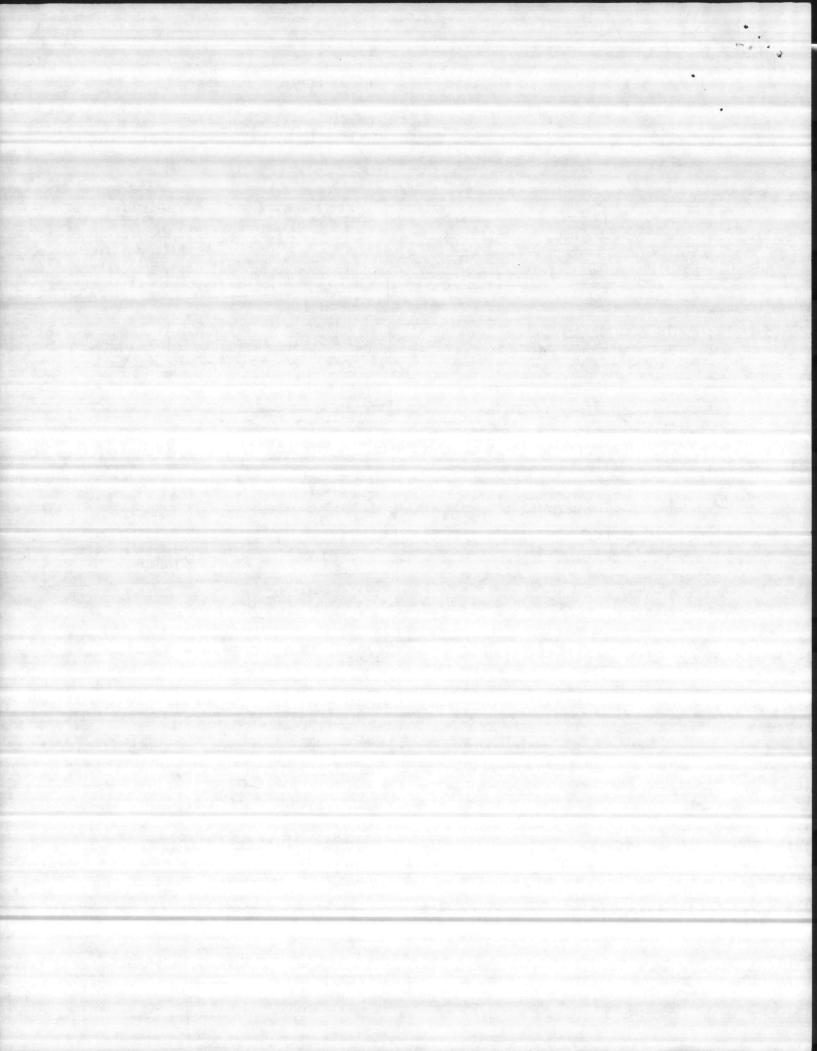
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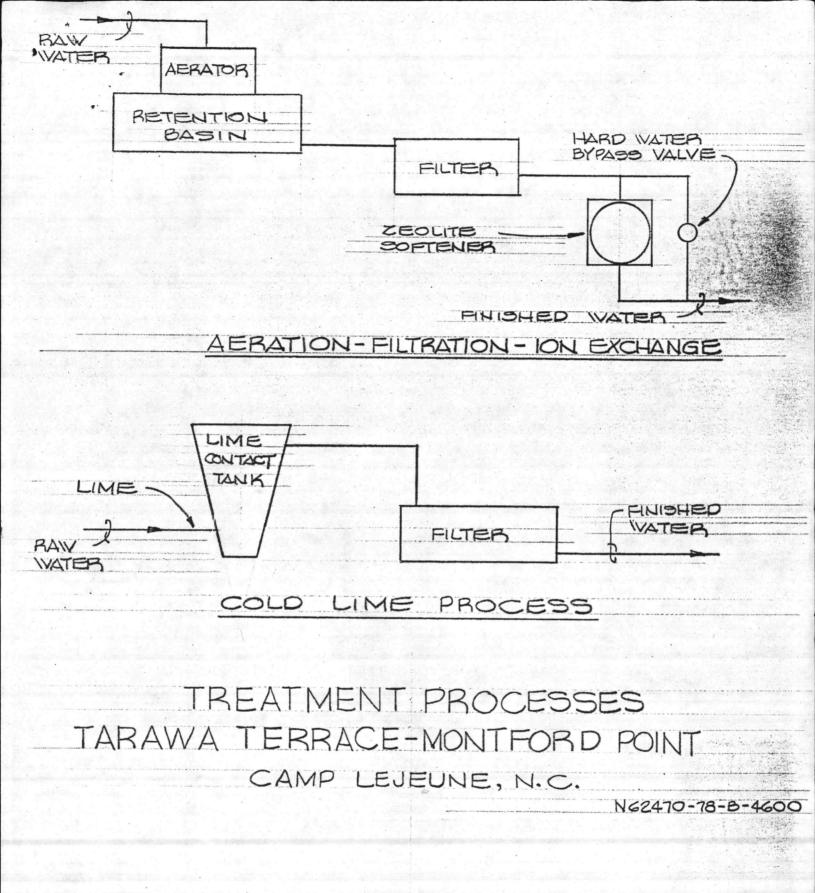




		ANA	LYSIS			
Color (000)		0 units	Ph	(00.0)		
		Results in Pa	rts per Millior		7.2	
Elkali-ity CaCO3	(000)	217				
Total Harsness	(000)			(0.00)	0.13	
-0- /-		282	Arsenic	(-0.00)	< 0.01	
	00.00)	0.23	Cadhium	(*0.00)	< 0.01	
	(50.00)	< 0.03	Chronica ¹⁶	(*0.03)		
S:02	(000)	2.5	Cooper		< 0.05	
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HENRY VON OESEN & ASSOCIATES GII PRINCESS ST. WILMINGTON, N.C.

